

Enriching Formative and Summative Assessment in Chemistry

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Central goal

- Describe a **guiding framework for enriching assessments** in chemistry and discuss what these enriched assessments reveal about our students' learning

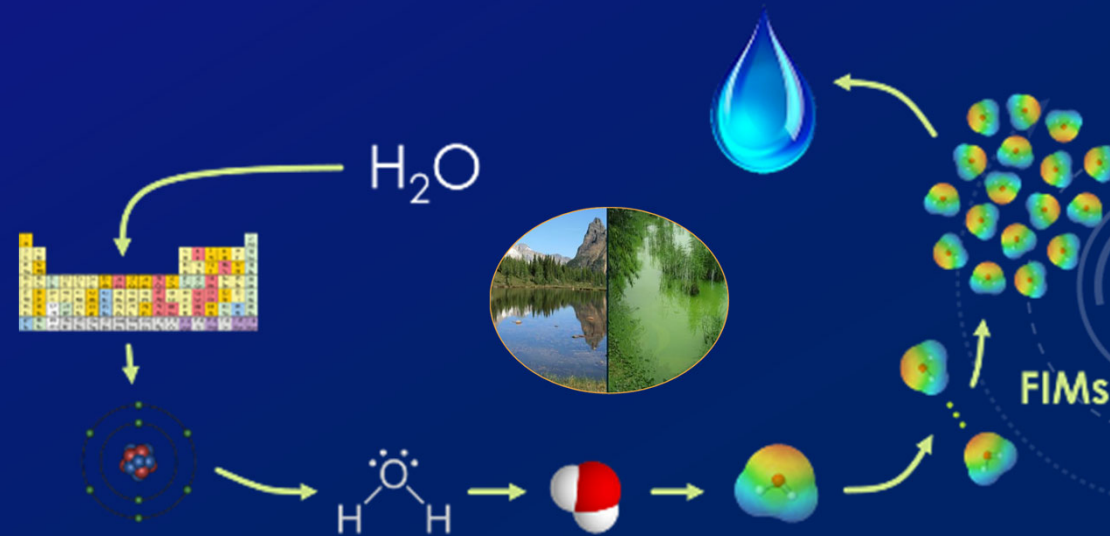
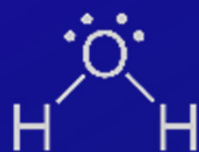


The challenge

FROM THE DEMONSTRATION
OF FRAGMENTED
KNOWLEDGE

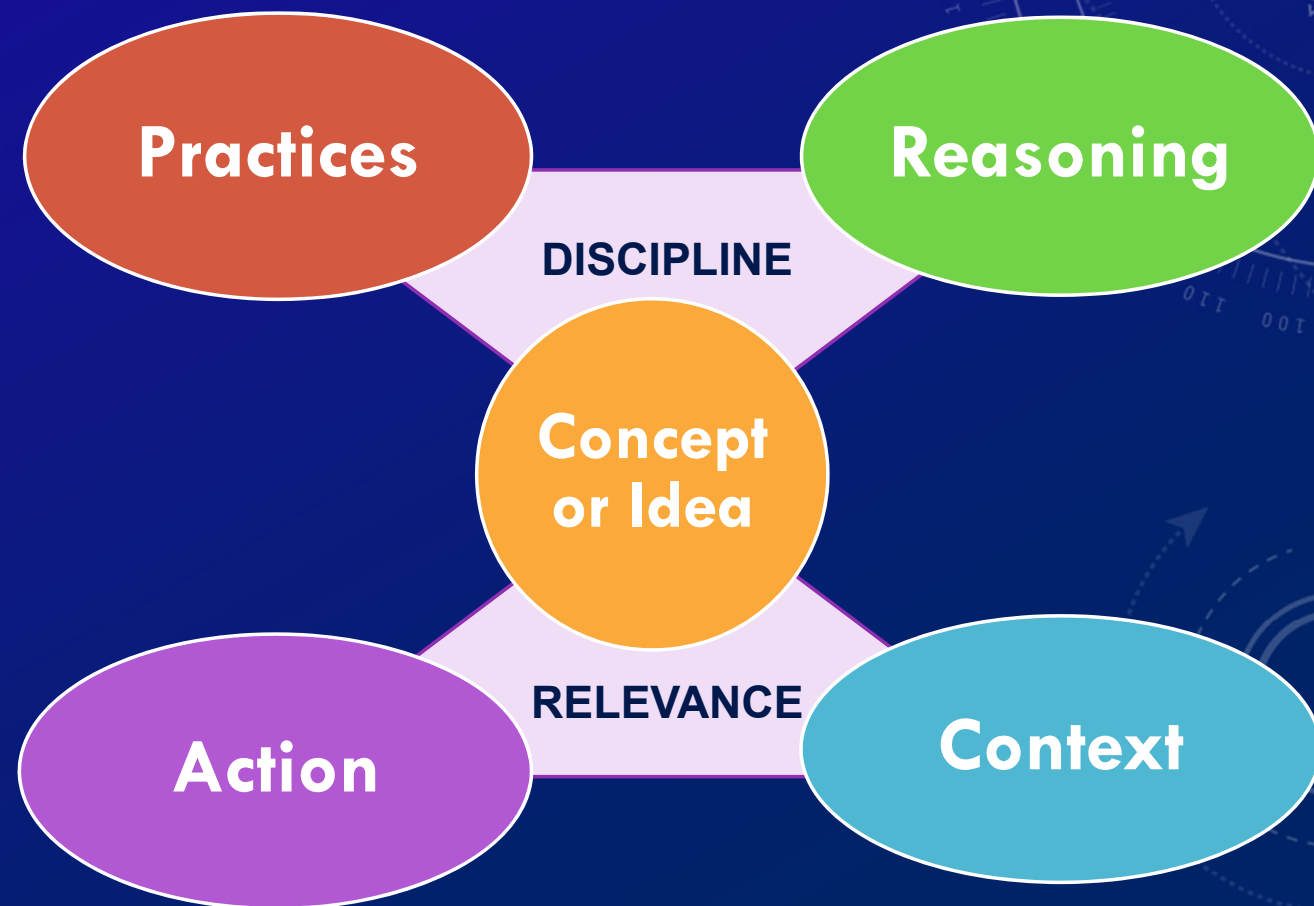


TO THE INTEGRATION AND
APPLICATION OF IDEAS
AND WAYS OF THINKING
IN RELEVANT CONTEXTS



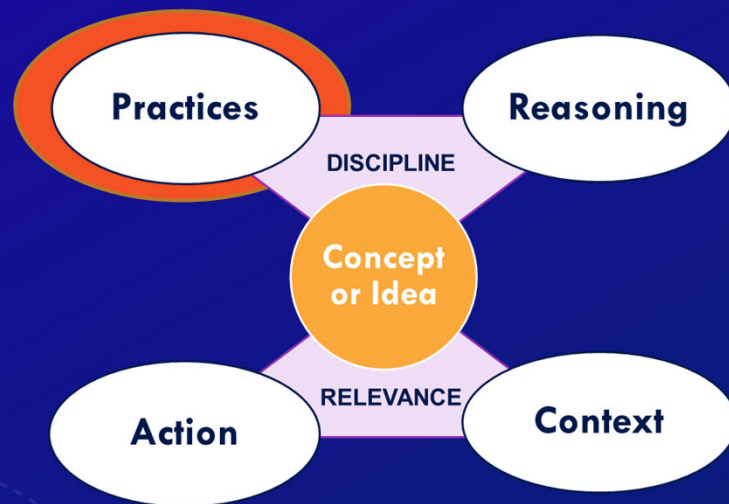
Guiding framework

Major dimensions for enrichment of chemistry assessments



The tradition

Traditional assessment questions target knowledge acquisition and comprehension

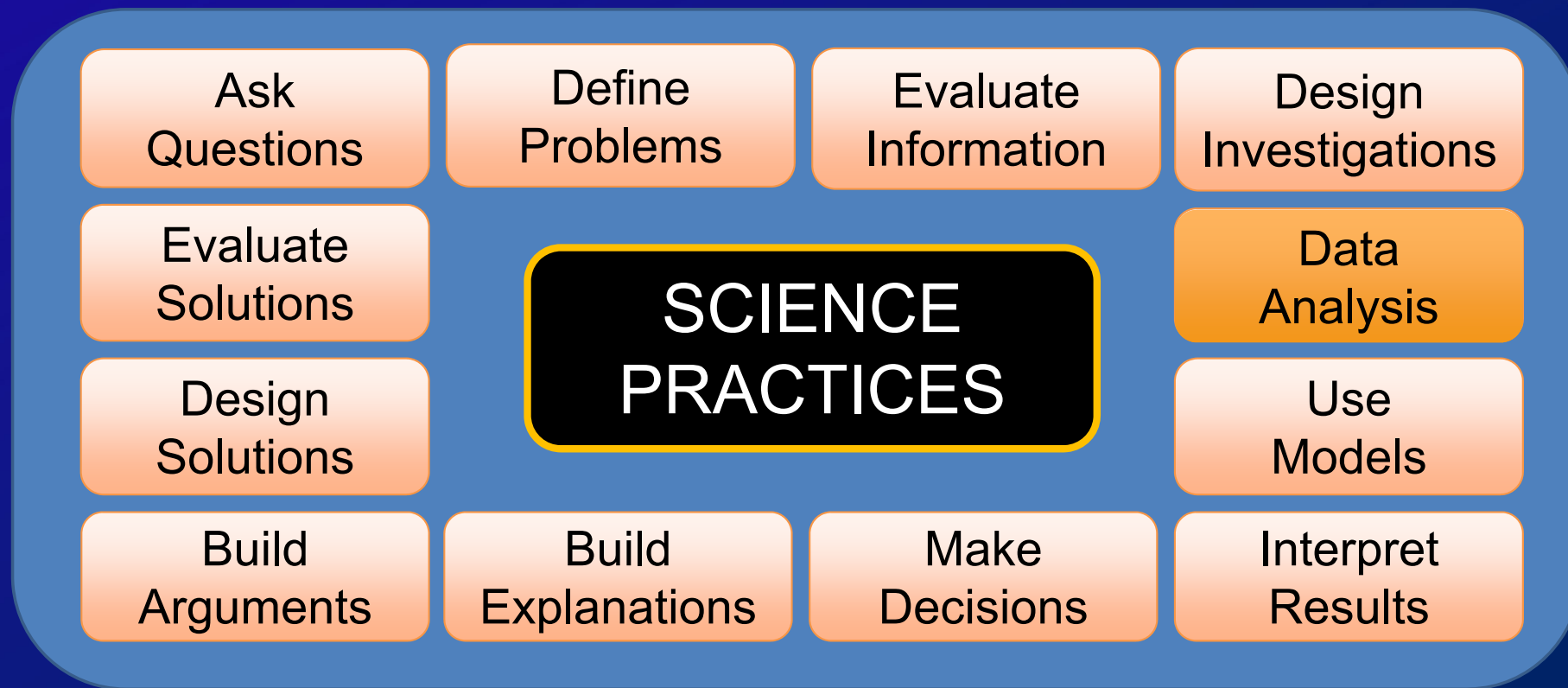


The solubility of a substance is a measure of:

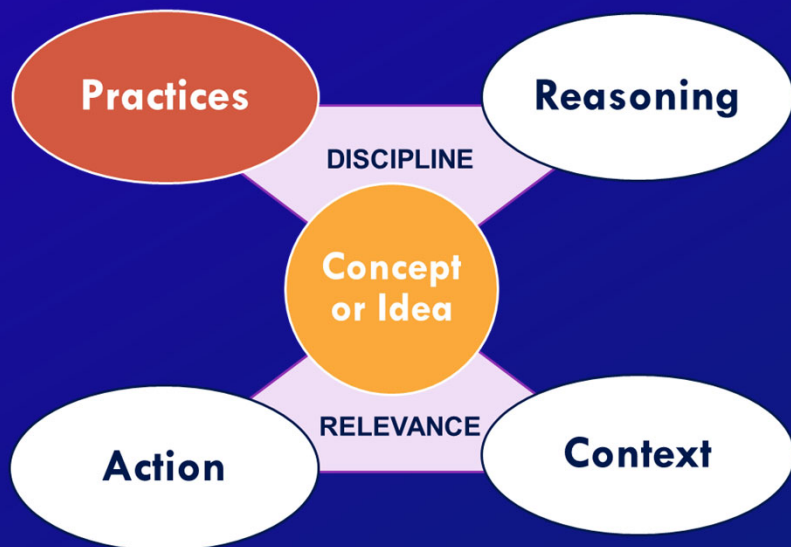
- A. The amount of substance in a solvent
- B. The maximum amount of substance that dissolves in a given volume of solvent
- C. The amount of substance that dissolves in one liter of solvent

Knowledge in action

Develop questions that assess students' ability to properly use knowledge when engaged in a science practice:

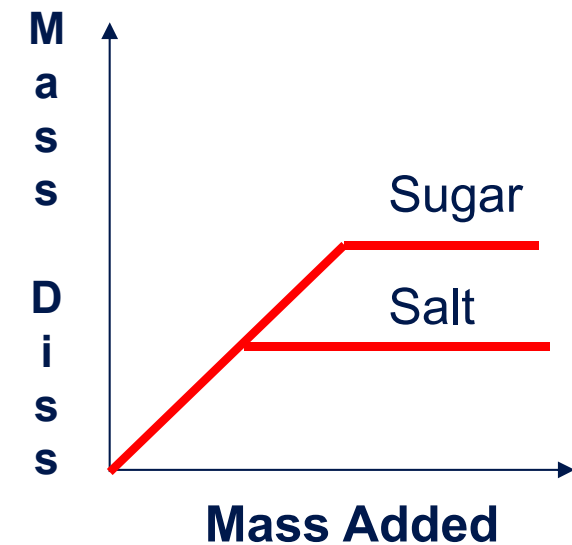


Practices dimension



The graph to the right represents the mass of sugar and salt dissolved in water as a function of the mass of each substance added to 100 mL of water. **Which of these substances is more soluble in water?**

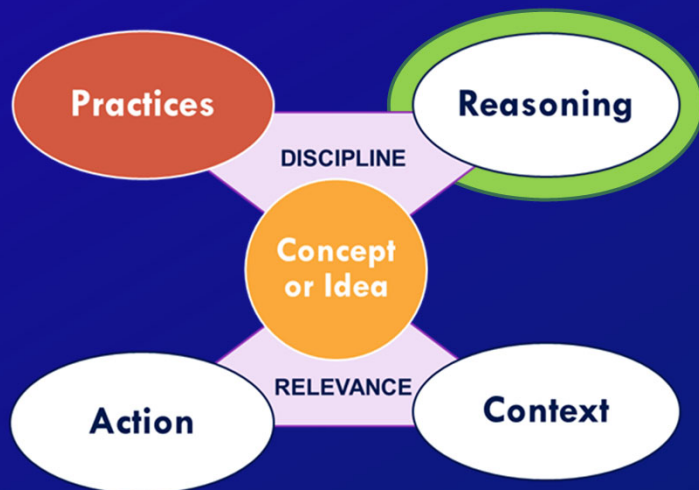
- A. Sugar
- B. Salt
- C. More information is needed



Data Analysis and Interpretation

The tradition

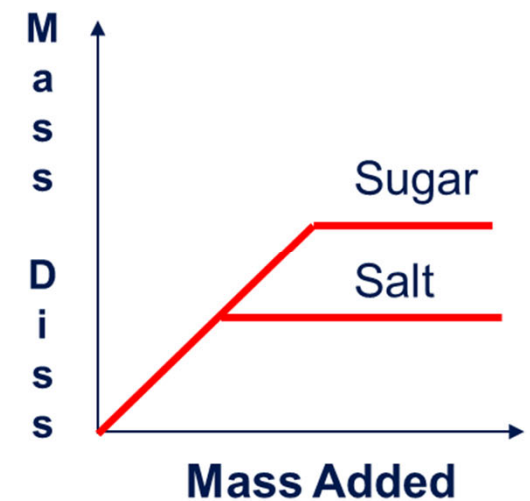
Traditional assessments typically evaluate if students know **WHAT** happens but not necessarily **WHY** it happens



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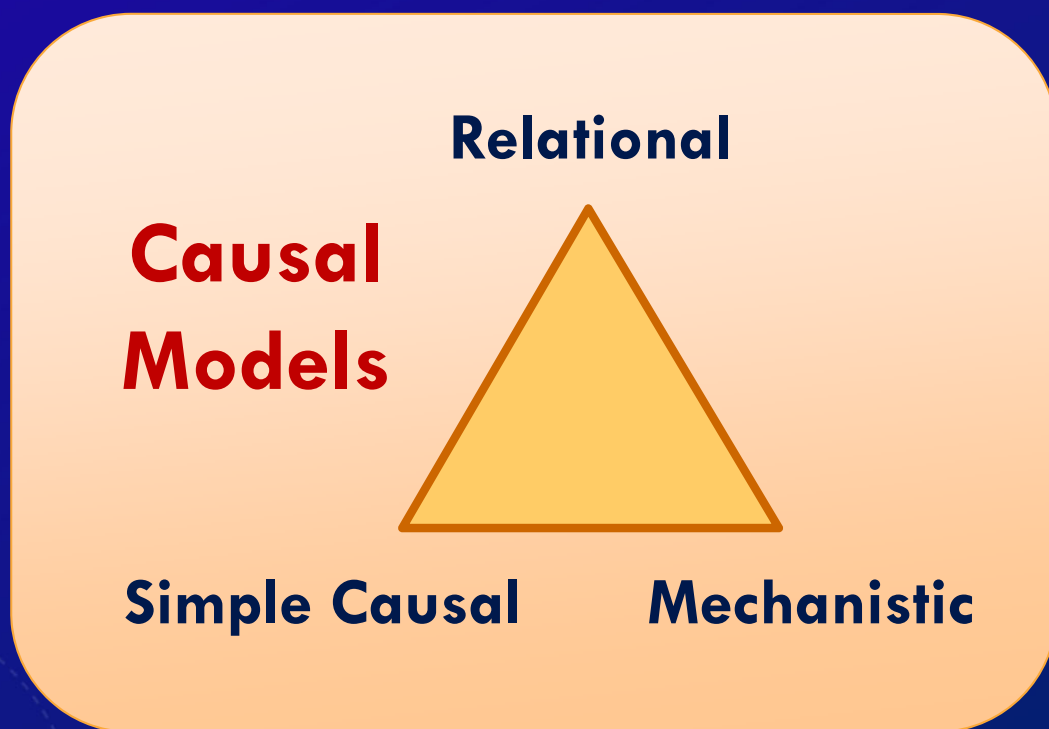
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Which? What?

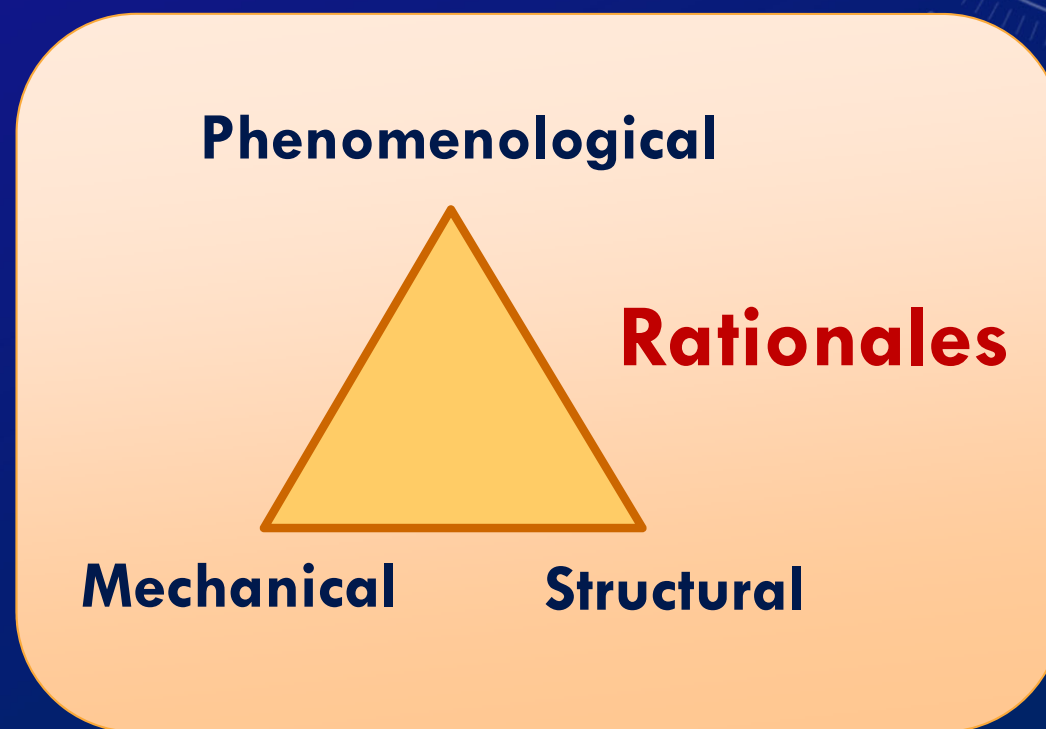


Chemical reasoning

Design questions that purposely require students to demonstrate different ways of reasoning in the discipline



Sevian & Talanquer, CERP, 2014, 15, 10-23



Talanquer, IJSE, 2018, 40, 1874-1890

Reasoning dimension

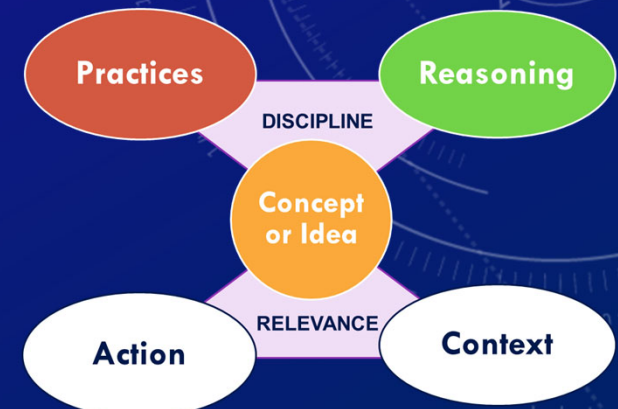
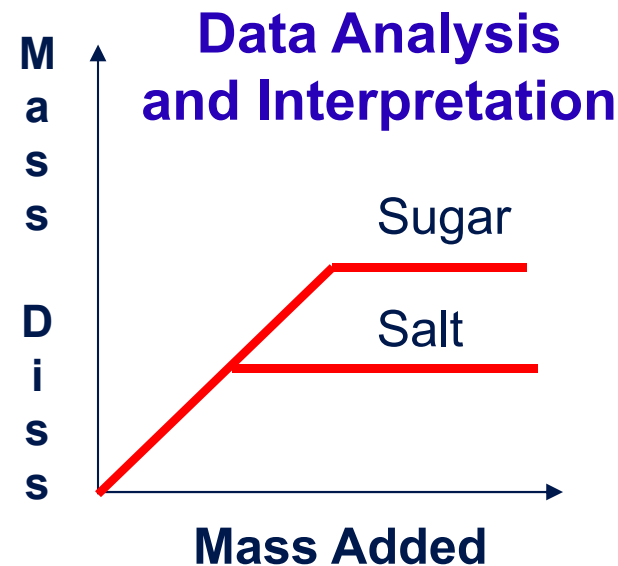
Laverty et al., PLOS. 2016, 0162333

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- A. Sugar
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Why? Explanation

- I. Because it takes more time for the salt to dissolve
- II. Because more sugar can be dissolved in the same volume of water
- III. Because more salt can be added to the same volume of water
- IV. Because we need to know the actual amounts that were used

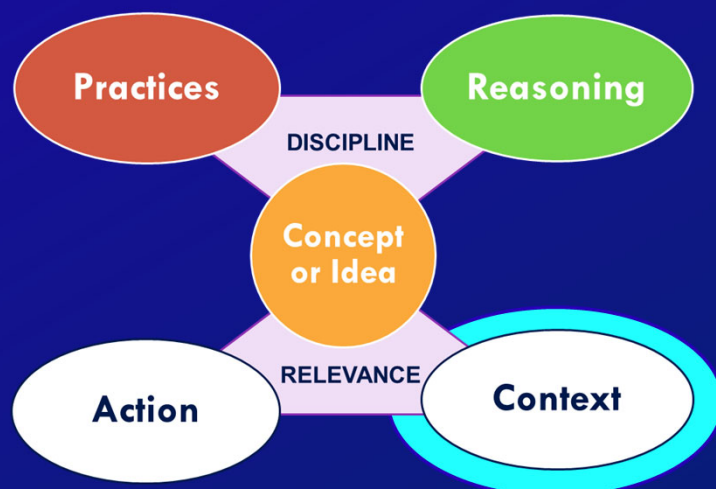


Phenomenological
Rationale

Relational

Academic contexts

Chemistry assessments tend to be set in academic rather than relevant contexts

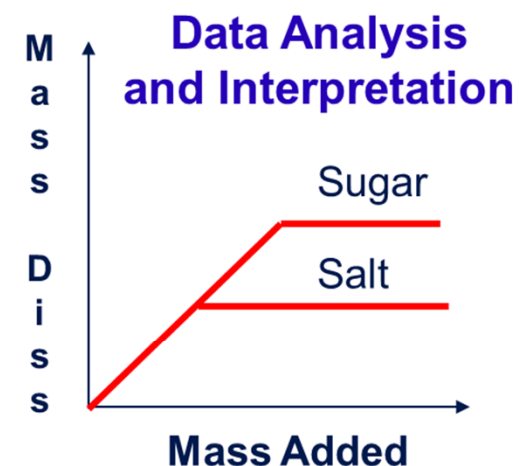


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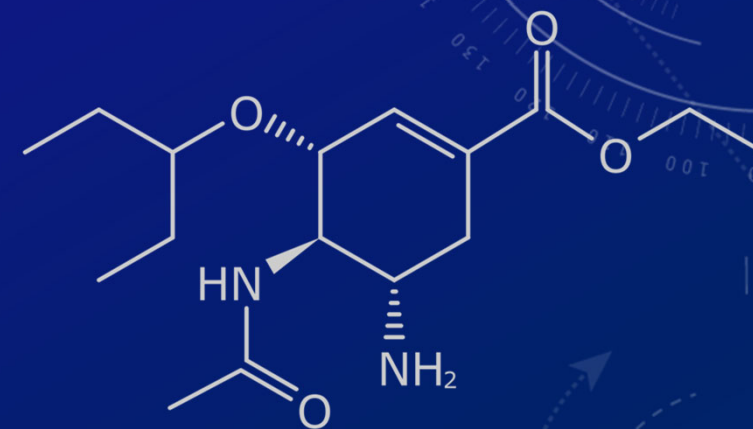
Contextual reasoning

Situated in relevant scenarios:

Broman, Bernholt, & Parchmann,
Res. Sci. Tech. Educ. **2015**, 33, 143–161

A medical drug against swine flu frequently discussed in media this year, Tamiflu (an antiviral agent), is unfortunately spreading in the environment. Explain how this can occur:

- What** features and properties do the molecules of this drug have that will make them spread in the environment?
- How** and **why** will those features and properties facilitate the spreading?



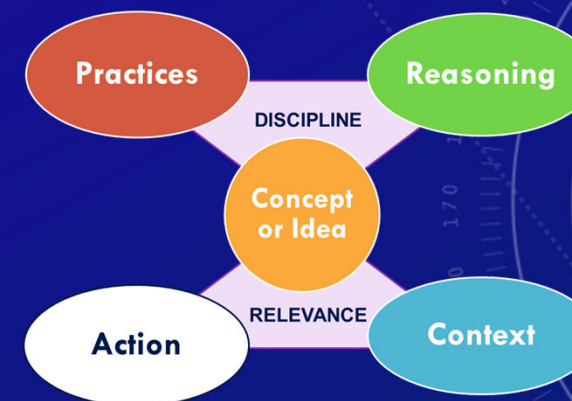
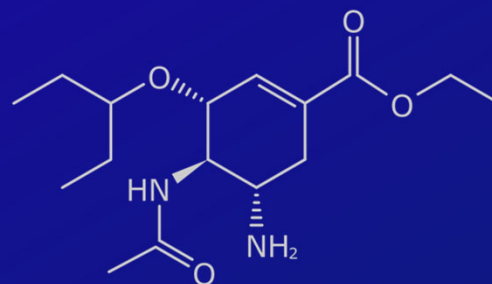
Structural Rationale

Mechanistic

Context dimension

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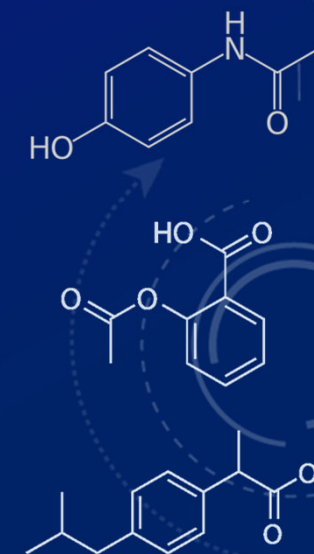
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Broman & Parchmann,
Chem. Educ. Res. Pract., 2014, 15, 516-529

When medical drugs are used in society, there are always effluents in the environment, partly from the surplus from drugs taken by humans, partly from industries when producing the drugs.

- What** features and properties do these types of molecules (e.g., Alvedon, Treo, Ipren) have that will make them spread in the environment?
- How** and **why** will those features and properties facilitate the spreading?



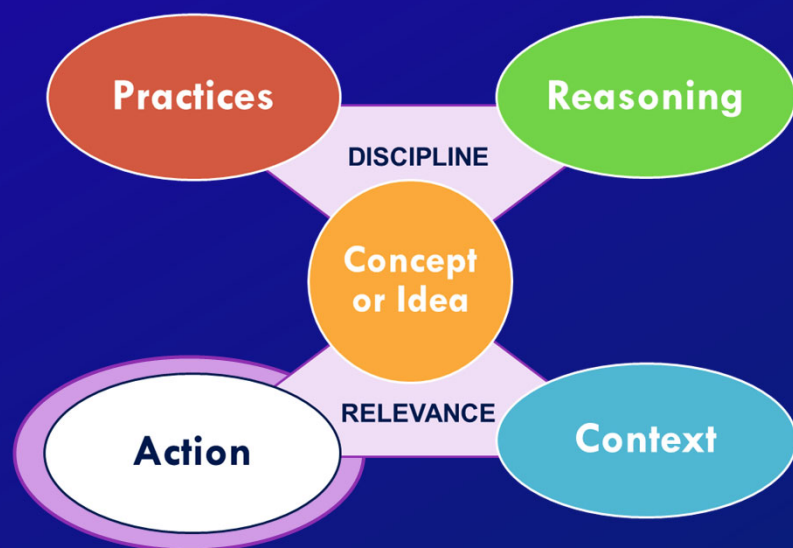
Specific



General

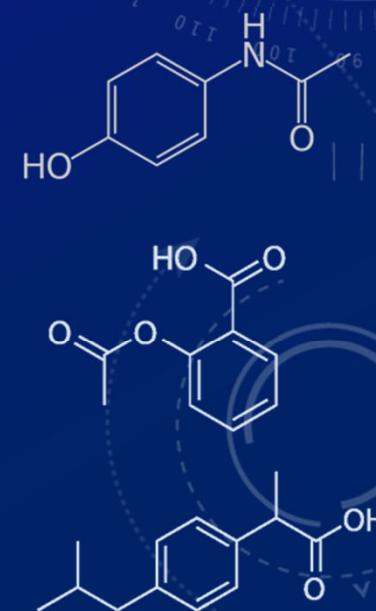
Socio-ecological reasoning

Chemistry assessments do not often engage students in decision-making based on benefit-cost-risk analyses



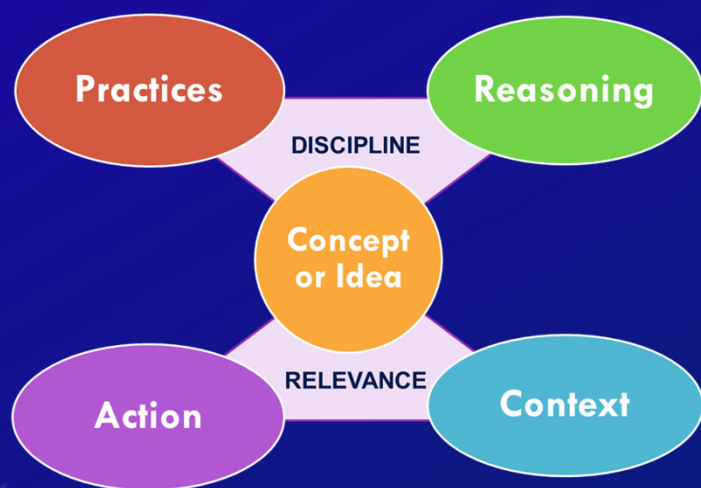
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- How** and **why** will those features and properties facilitate the spreading?



Action dimension

Require students to apply their knowledge to evaluate options and make decisions:



In a battery factory workers are exposed to ZnS and CdCl_2 , HCl , oily grease, CH_2Cl_2 , and H_2S . A suggestion was made to replace water with petroleum for washing the employers' work clothes.

- Identify benefits-costs-risks** of replacing the water with petroleum from the point of view of cleaning the clothes and of the environment outside the factory. **Justify your claims based on the composition and structure** of the substances involved
- Use the results of your analysis to **make a recommendation** about the suggested replacement and clearly justify your reasoning

Chemical thinking curriculum

Talanquer, & Pollard
CERP, 2010, 11, 74-83
J. Chem. Educ. 2017, 94, 1844-1851

What are its
impacts?

What is this
made of?

What properties
does it have?



How can we
control it?

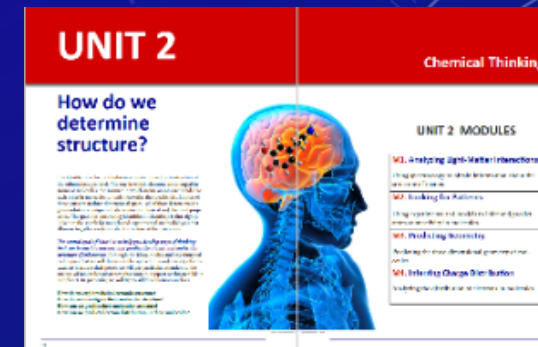
Why does it
happen?

How does it
happen?

<https://sites.google.com/site/chemicalthinking/>

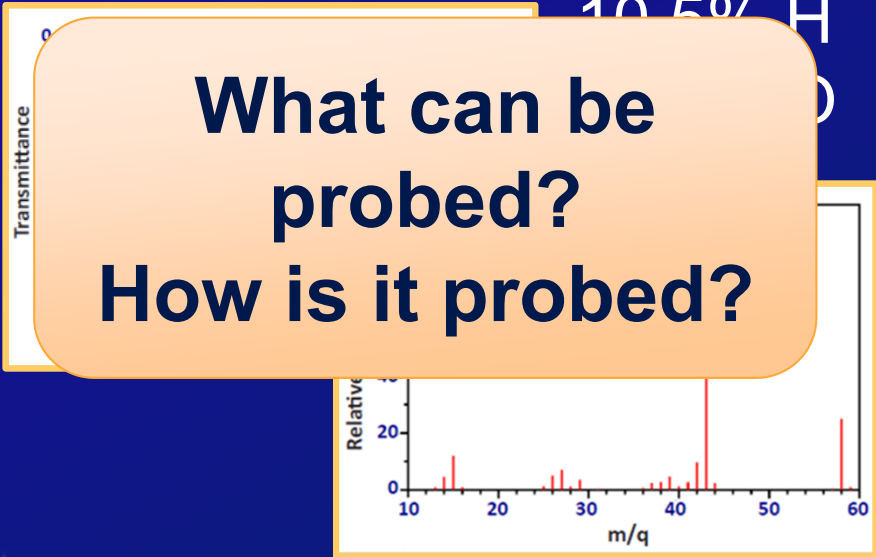
Instructional tasks

What is in your breath?



DATA ANALYSIS

MODELING



What can be probed?
How is it probed?

What can be inferred and how?
How is it justified?

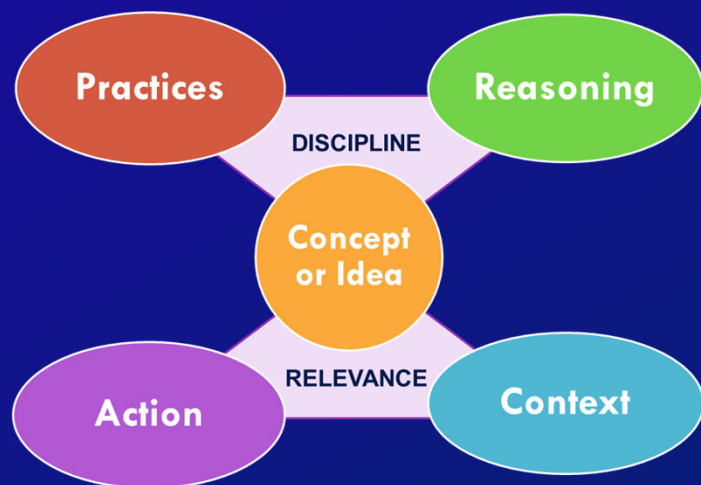
ARGUMENTATION

62.0% C
10.5% H
D

C H O

Scenario-based assessments

Used both in the design of formative and summative assessments



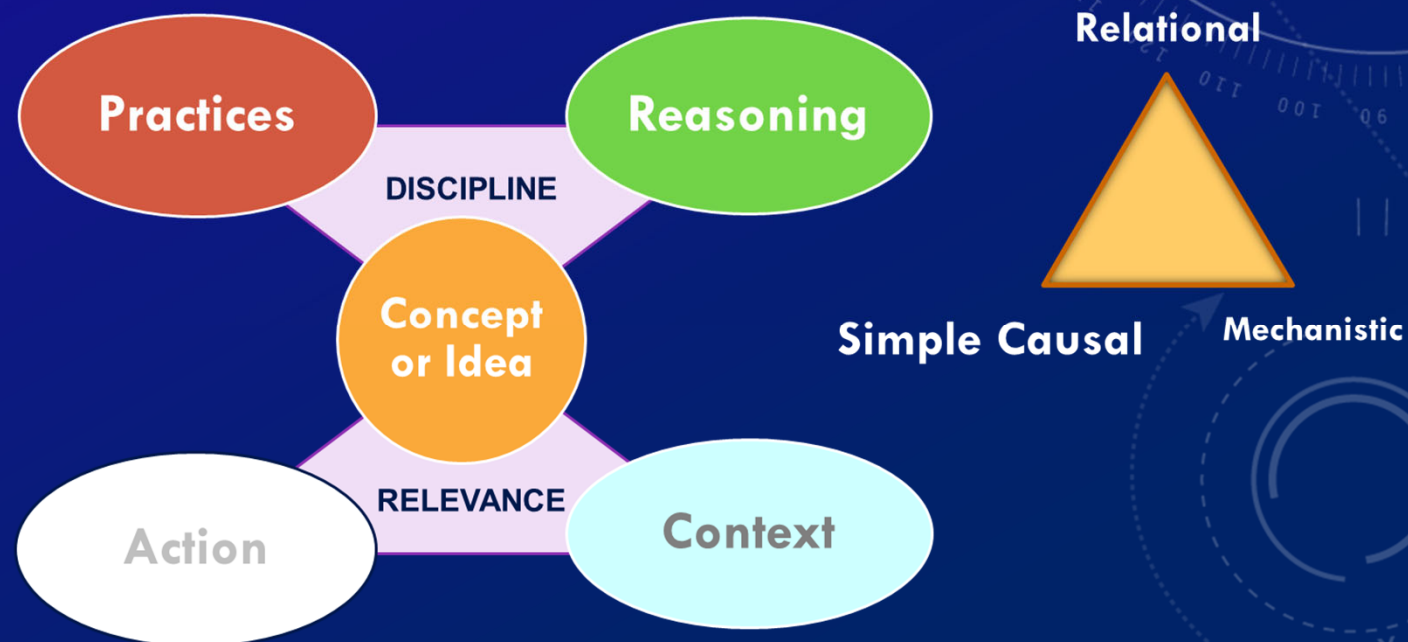
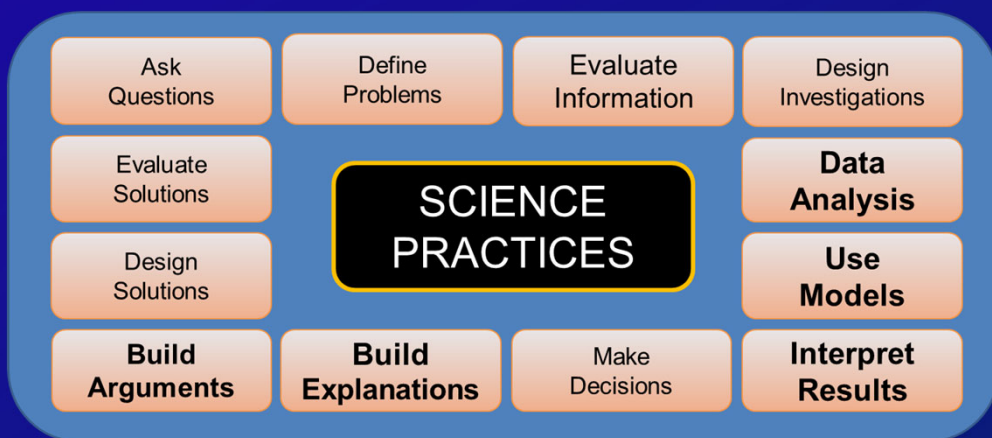
The image displays four scenario-based assessment cards, each with a title and detailed content:

- Food Additives:** Discusses food additives, their uses, and safety. Includes sections for Citric Acid, Phosphoric Acid, and Aspartame, with chemical structures and safety information.
- Psychotropics:** Discusses psychoactive drugs, their effects, and safety. Includes sections for LSD, Amphetamines, and Cocaine, with chemical structures and safety information.
- Acids in the Air:** Discusses air pollution, acid rain, and its effects. Includes sections for Sulfuric Acid, Nitric Acid, and Chloric Acid, with chemical structures and safety information.
- Proteins:** Discusses protein structure, function, and denaturation. Includes sections for Denatured Protein and Amino Acid Structure, with chemical structures and safety information.

<https://sites.google.com/site/chemicalthinking/>

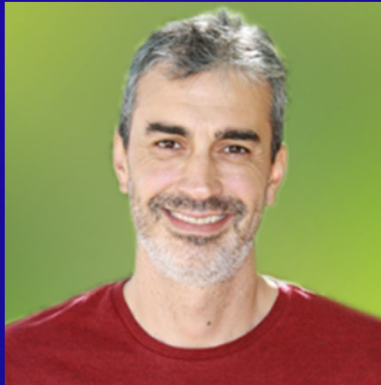
Strengths and weaknesses

The framework is also useful in the educational impact of our program



Talanquer, *J. Chem. Educ.*, 2019, 96, 2918–2925

Acknowledgments



John Pollard



Adam
Daly



Amy
Graham



Tori
Hidalgo



Suchi
Perera



Brian
Zacher



Mark Yanagihashi



Thank You

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<https://sites.google.com/site/chemicalthinking/>



Questions? Comments?